

## MATERIAL SAFETY DATA SHEET

PRODUCT NAME: CARBON DIOXIDE, GAS

# 1. Product and Company Identification

BOC Gases,
Division of,
BOC Gases
Division of

The BOC Group, Inc.

BOC Canada Limited

575 Mountain Avenue 5975 Falbourne Street, Unit 2 Murray Hill, NJ 07974 Mississauga, Ontario L5R 3W6

**TELEPHONE NUMBER:** (908) 464-8100 **TELEPHONE NUMBER:** (905) 501-1700

24-HOUR EMERGENCY TELEPHONE NUMBER: 24-HOUR EMERGENCY TELEPHONE NUMBER:

CHEMTREC (800) 424-9300 (905) 501-0802

**EMERGENCY RESPONSE PLAN NO: 2-0101** 

PRODUCT NAME: CARBON DIOXIDE, GAS

CHEMICAL NAME: Carbon Dioxide

COMMON NAMES/SYNONYMS: Carbonic Anhydride

TDG (Canada) CLASSIFICATION: 2.2

WHMIS CLASSIFICATION: A

PREPARED BY: Loss Control (908)464-8100/(905)501-1700

**PREPARATION DATE:** 6/1/95 **REVIEW DATES:** 11/18/03

# 2. Composition, Information on Ingredients

#### EXPOSURE LIMITS<sup>1</sup>:

INGREDIENT	% VOLUME	PEL-OSHA <sup>2</sup>	TLV-ACGIH <sup>3</sup>	LD <sub>50</sub> or LC <sub>50</sub> Route/Species
Carbon Dioxide FORMULA: CO <sub>2</sub> CAS: 124-38-9 RTECS #: FF6400000	99.8 TO 99.999	5000 ppm TWA	5000 ppm TWA 30,000 ppm STEL	Not Available

<sup>&</sup>lt;sup>1</sup> Refer to individual state or provincial regulations, as applicable, for limits which may be more stringent than those listed here.

IDLH (Carbon Dioxide): 40,000 ppm

OSHA Regulatory Status: This material is classified as hazardous under OSHA regulations.

# 3. Hazards Identification

# **EMERGENCY OVERVIEW**

Odorless, colorless, nonflammable gas. Simple Asphyxiant – This product does not contain oxygen and may cause asphyxia if released in a confined area. Maintain oxygen levels above 19.5%. Carbon dioxide acts as a weak narcotic at high concentrations (30,000 ppm). Inhalation of high concentrations of carbon dioxide can cause reduced hearing acuity, changes in respiration and increased blood pressure and pulse. Contents under pressure. Use and store below 125 °F.

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<sup>&</sup>lt;sup>2</sup> As stated in 29 CFR 1910, Subpart Z (revised July 1, 1993)

<sup>&</sup>lt;sup>3</sup> As stated in the ACGIH 2003 Threshold Limit Values for Chemical Substances and Physical Agents.

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#### **ROUTE OF ENTRY:**

Skin Contact	Skin Absorption	Eye Contact	Inhalation	Ingestion
No	No	No	Yes	No

#### **HEALTH EFFECTS:**

Exposure Limits	Irritant	Sensitization
Yes	No	No
Teratogen	Reproductive Hazard	Mutagen
No	No	No
Synergistic Effects		
None reported		

Carcinogenicity: -- NTP: No IARC: No OSHA: No

#### **EYE EFFECTS:**

Liquid and cold vapor may cause tissue freezing.

#### **SKIN EFFECTS:**

May cause frostbite.

#### **INGESTION EFFECTS:**

No adverse effects anticipated.

#### **INHALATION EFFECTS:**

Depending on concentration and duration of exposure carbon dioxide may cause increased respiration, headache, mild narcotic effects, increased blood pressure and pulse, and asphyxiation. Symptoms of overexposure to carbon dioxide become more apparent when atmospheric oxygen is decreased to 15-17%. Chronic harmful effects are not known from repeated inhalation of concentrations below the PEL/TLV.

#### MEDICAL CONDITIONS AGGRAVATED BY EXPOSURE: None known.

POTENTIAL ENVIRONMENTAL EFFECTS: Not expected to be toxic to fish and wildlife.

## 4. First Aid Measures

#### **EYES:**

For liquid or cold vapor, flush eyes with lukewarm water and obtain immediate medical attention.

#### SKIN:

For frostbite, immediately warm affected area with lukewarm water (< 105 °F).

# **INGESTION:**

Not anticipated.

#### **INHALATION:**

PROMPT MEDICAL ATTENTION IS MANDATORY IN ALL CASES OF OVEREXPOSURE TO CARBON DIOXIDE. RESCUE PERSONNEL SHOULD BE EQUIPPED WITH SELF-CONTAINED BREATHING APPARATUS. Conscious persons should be assisted to an uncontaminated area and inhale fresh air. Quick removal from the contaminated area is most important. Unconscious persons should be moved to an uncontaminated area, given mouth-to-mouth resuscitation and supplemental oxygen. Further treatment should be symptomatic and supportive.

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# 5. Fire Fighting Measures

Conditions of Flammability: Nonflammable			
Flash point:	Method:		Autoignition
None	Not Applicable		Temperature: None
LEL(%): None		UEL(%): None	
Hazardous combustion products: None			
Sensitivity to mechanical shock: None			
Sensitivity to static discharge: None			

#### FIRE AND EXPLOSION HAZARDS:

Nonflammable. Cylinder may vent rapidly or rupture violently from pressure when involved in a fire situation.

#### **EXTINGUISHING MEDIA:**

None required. Use media appropriate for surrounding fire.

#### FIRE FIGHTING INSTRUCTIONS:

Firefighters should wear respiratory protection (SCBA) and full turnout or Bunker gear. Continue to cool fire-exposed containers until well after flames are extinguished.

#### 6. Accidental Release Measures

Evacuate all personnel from affected area. Use appropriate protective equipment (See Section 8). Stop the flow of gas or remove cylinder to outdoor location if this can be done without risk. Ventilate enclosed areas. If leak is in user's equipment, be certain to purge piping with inert gas prior to attempting repairs. If leak is in container or container valve, contact the appropriate emergency telephone number listed in Section 1 or call your closest BOC location.

# 7. Handling and Storage

Electrical Classification: Non-Hazardous

Dry carbon dioxide can be handled in most common structural materials. Moist carbon dioxide is generally corrosive by its formation of carbonic acid. For applications with moist Carbon Dioxide, 316, 309 and 310 stainless steels may be used as well as Hastelloy <sup>®</sup> A, B, & C, and Monel <sup>®</sup>. Ferrous Nickel alloys are slightly susceptible to corrosion. At normal temperatures carbon dioxide is compatible with most plastics and elastomers.

Use only in well-ventilated areas. Carbon dioxide vapor is heavier than air and will accumulate in low areas. Valve protection caps must remain in place unless container is secured with valve outlet piped to use point. Do not drag, slide or roll cylinders. Use a suitable hand truck for cylinder movement. Use a pressure reducing regulator when connecting cylinder to lower pressure piping or systems. Do not heat cylinder by any means to increase the discharge rate of product from the cylinder. Use a check valve or trap in the discharge line to prevent hazardous back flow into the system. Do not insert any object (i.e.: screwdriver) into valve cap openings as this can damage the valve causing leakage.

Protect cylinders from physical damage. Store in cool, dry, well-ventilated area away from heavily trafficked areas and emergency exits. Do not allow the temperature where cylinders are stored to exceed 125°F (52°C). Cylinders should be stored upright and firmly secured to prevent falling or being knocked over. Full and empty cylinders should be segregated. Use a "first in-first out" inventory system to prevent full cylinders being stored for excessive periods of time.

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For additional storage recommendations, consult Compressed Gas Association's Pamphlet P-1, AV-7, G-6, G-6.1, G-6.2, G-6.3, G-6.5, G-6.7, G-6.9, PS-5 and TB-10.

Never carry a compressed gas cylinder or a container of a gas in cryogenic liquid form in an enclosed space such as a car trunk, van or station wagon. A leak can result in a fire, explosion, asphyxiation or a toxic exposure.

# 8. Exposure Controls, Personal Protection

#### **ENGINEERING CONTROLS:**

Use local exhaust in combination with general ventilation as necessary to control air contaminants to at or below acceptable exposure guidelines.

**EYE/FACE PROTECTION:** Safety goggles or glasses as appropriate for the job.

**SKIN PROTECTION:** Protective gloves of any material appropriate for the job.

#### RESPIRATORY PROTECTION:

For emergency release use a positive pressure NIOSH approved air-supplying respirator systems (SCBA or airline/escape bottle) using at a minimum Grade D air.

#### OTHER/GENERAL PROTECTION:

Safety shoes.

# 9. Physical and Chemical Properties

PARAMETER	VALUE	UNITS
Physical state (gas, liquid, solid)	: Gas	
Vapor pressure at 70 °F	: 856	psia
Vapor density at 70 °F, 1 atm (Air = 1)	: 1.53	
Evaporation point	: Not Available	
Boiling point (CO2 Sublimes)	: -109.3	°F
	: -78.5	$^{\circ}\mathrm{C}$
Freezing point	: -69.8	°F
	: -56.6	$^{\circ}\mathrm{C}$
pH	: Not Available	
Specific gravity	: Not Available	
Oil/water partition coefficient	: Not Available	
Solubility (H <sub>2</sub> 0)	: High, 0.145	g/ml, @25 °C
Odor threshold	: Not Applicable	
Odor and appearance	: A colorless, odorless	gas.

# 10. Stability and Reactivity

#### **STABILITY:**

Stable

#### INCOMPATIBLE MATERIALS/CONDITIONS:

Certain reactive metals, hydrides, moist cesium monoxide, or lithium acetylene carbide diammino may ignite. Passing carbon dioxide over a mixture of sodium peroxide and aluminum or magnesium may explode.

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#### **HAZARDOUS DECOMPOSITION PRODUCTS:**

Carbon monoxide and oxygen when heated above 3092 °F (1700°C). Carbonic acid is formed in the presence of moisture.

#### HAZARDOUS POLYMERIZATION:

Will not occur.

# 11. Toxicological Information

**SKIN AND EYE:** Adverse effects are not expected.

**INHALATION:** Acidosis, adrenal cortical exhaustion, and other metabolic stresses have resulted from prolonged continuous exposure to 1-2% carbon dioxide (10,000 ppm – 20,000 ppm). The ACGIH TLV of 5,000 ppm is expected to provide a good margin of safety from asphyxiation and undue metabolic stress provided sufficient oxygen levels are maintained in the air. Increased physical activity, duration of exposure, and decreased oxygen content can affect systemic and respiratory effects resulting from exposure to carbon dioxide.

**OTHER:** Oxygen deficiency during pregnancy has produced developmental abnormalities in humans and experimental animals.

Exposure of female rats to 60,000 ppm carbon dioxide for 24 hours has produced toxic effects to the embryo and fetus in pregnant rats. Toxic effects to the reproductive system have been observed in other mammalian species at similar concentrations.

Chronic, harmful effects are not known from repeated inhalation of low (3 to 5 molar %) concentrations.

# 12. Ecological Information

Product does not contain Class I or Class II ozone depleting substances. Not toxic. Will not bioconcentrate.

#### 13. Disposal Considerations

Do not attempt to dispose of residual waste or unused quantities. Return in the shipping container PROPERLY LABELED, WITH ANY VALVE OUTLET PLUGS OR CAPS SECURED AND VALVE PROTECTION CAP IN PLACE to BOC Gases or authorized distributor for proper disposal.

# 14. Transport Information

PARAMETER	United States DOT	Canada TDG
PROPER SHIPPING NAME:	Carbon Dioxide	Carbon Dioxide
HAZARD CLASS:	2.2	2.2
IDENTIFICATION NUMBER:	UN 1013	UN 1013
SHIPPING LABEL:	NONFLAMMABLE GAS	NONFLAMMABLE GAS

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# 15. Regulatory Information

# SARA TITLE III NOTIFICATIONS AND INFORMATION SARA TITLE III HAZARD CLASSES:

Acute Health Hazard.

Sudden Release of Pressure Hazard

#### SARA TITLE III - SECTION 313 SUPPLIER NOTIFICATION:

This product does not contain toxic chemicals subject to reporting requirements of section 313 of the Emergency Planning and Community Right-To-Know Act (EPCRA) of 1986 and of 40 CFR 372.

**U.S. TSCA/Canadian DSL:** All ingredients are listed on the U.S. Toxic Substances Control Act (TSCA) inventory or exempt from listing and on the Canadian Domestic Substance List (DSL).

**California Proposition 65:** This product does not contain ingredient(s) known to the State of California to cause cancer or reproductive toxicity.

**Canadian Controlled Products Regulations (CPR):** This product has been classified in accordance with the hazard criteria of the CPR and the MSDS contains all the information required by the CPR.

#### 16. Other Information

# Health: 1 Health: 0 0 = No Hazard Flammability: 0 Flammability: 0 1 = Slight Hazard Instability: 0 Reactivity: 0 2 = Moderate Hazard 3 = Serious Hazard 4 = Severe Hazard

Note: The Reactivity Hazard Rating is based on the 2<sup>nd</sup> Edition of the National Paint and Coatings Association's (NPCA's) Hazardous Materials Identification System (HMIS<sup>®</sup>). Hazard ratings were based on the best available information at the time of the review. Ratings will be reassigned in accordance with Compressed Gas Association (CGA) guidelines as published in the future edition of CGA Pamphlet P-19.

ACGIH	American Conference of Governmental Industrial Hygienists
DOT	Department of Transportation

DOT Department of Transportation

IARC International Agency for Research on Cancer

NTP National Toxicology Program

OSHA Occupational Safety and Health Administration

PEL Permissible Exposure Limit

SARA Superfund Amendments and Reauthorization Act

STEL Short Term Exposure Limit

TDG Transportation of Dangerous Goods

TLV Threshold Limit Value

WHMIS Workplace Hazardous Materials Information System

Compressed gas cylinders shall not be refilled without the express written permission of the owner. Shipment of a compressed gas cylinder which has not been filled by the owner or with his/her (written) consent is a violation of transportation regulations.

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